

Fermilab

p̄ Note #143  
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NOTES ON BEAM PROFILE MONITOR

D. Joutras\*

M. Reinhardt\*

\*University of Wisconsin

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## Magnesium Jet

The design goals of the magnesium jet were to 1) reach an ion count rate of greater than 10,000 cps, 2) keep the detector system from being contaminated with magnesium, and 3) have a long operating time ( $> 1000$  hrs) without magnesium replacement.

### Assumed Beam and Vacuum Characteristics

$10^7$  protons circulating at 1 Mhz

$$\Rightarrow N_i \text{ (number of protons passing given point)} \\ = 10^{13} \text{ protons/sec}$$

$$\sigma_m \text{ (ionization cross section for protons on magnesium)} \\ = 2 \times 10^{-18} \text{ cm}^2$$

$$\text{background pressure in beam pipe} = 5 \times 10^{-10} \text{ T}$$

$$\text{background counts} \sim 500 \text{ cps}$$

### Oven Characteristics

(Oven design based on RADAX II specifications.)

$$\text{effective crucible capacity} = 8 \text{ cm}^3$$

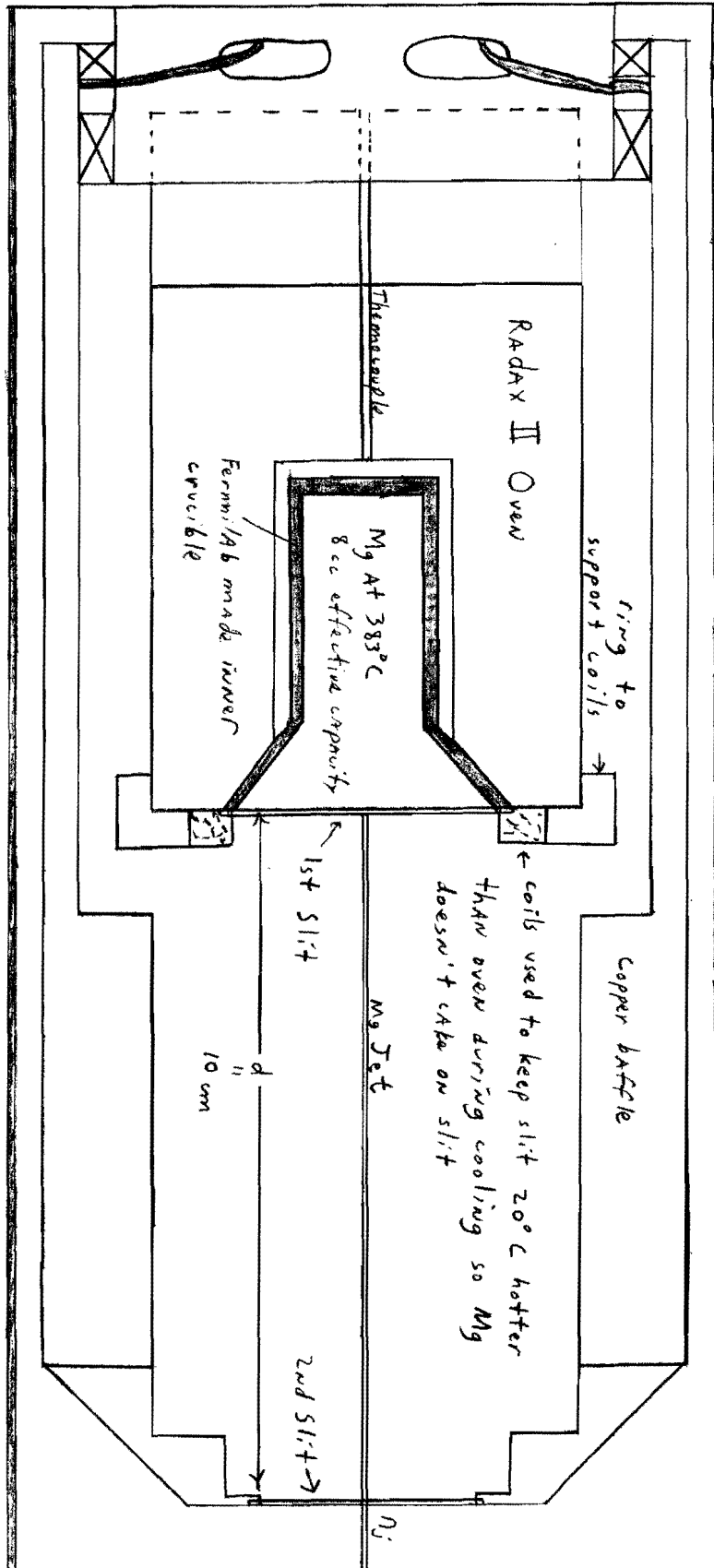
$$\text{operating temp.} = 383^\circ \text{C} \quad T = 656^\circ \text{K}$$

$$n_0 \text{ (density of magnesium leaving oven)} = 1.47 \times 10^{13} \frac{\text{part}}{\text{cm}^3}$$

Fig 1 : Oven Assembly

Slit 2:  $3\text{ cm} \times 1\text{ mm}$  Area  $A_2 = 3 \times 10^{-1}\text{ cm}^2$

$n_j$  = Jet density



oven slit dimensions:

slit 1 60 .2 mm diameter holes,  
.5 mm center-to-center

$$v = \text{velocity of magnesium} = 14,551 \sqrt{\frac{T(K^\circ)}{m \text{ (g/mole)}}} \text{ cm/s}$$

$$m = 24.3 \text{ g/mole} \Rightarrow v = 7.56 \times 10^4 \frac{\text{cm}}{\text{s}}$$

### Mg Depletion

$$F_0 = \text{oven flow} = n_0 v A_1$$

$A_1 =$  area of slit #1

$$= 60 \pi r^2 = 1.885 \times 10^{-2} \text{ cm}^2$$

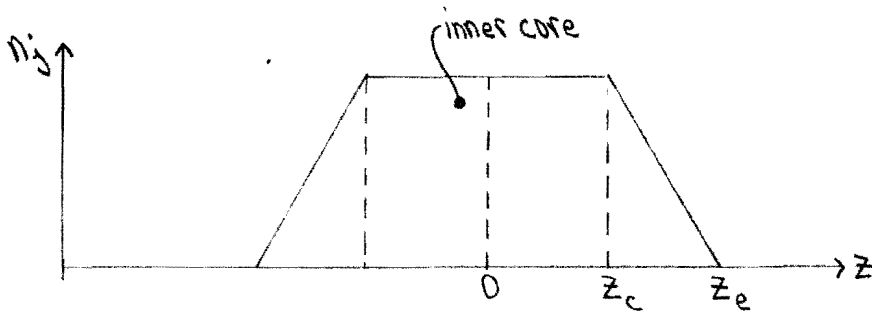
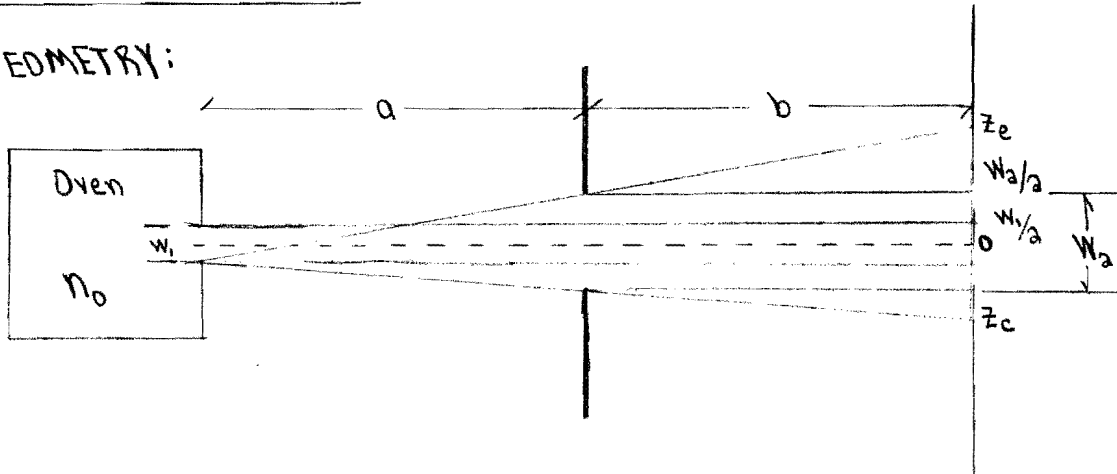
$$\begin{aligned} \therefore F_0 &= \left( 1.47 \times 10^{13} \frac{\text{part.}}{\text{cm}^3} \right) \left( 7.56 \times 10^4 \frac{\text{cm}}{\text{s}} \right) \left( 1.885 \times 10^{-2} \text{ cm}^2 \right) \\ &= 2.1 \times 10^{16} \frac{\text{part.}}{\text{sec}} \end{aligned}$$

$$\frac{F_0}{N_A} = \frac{2.1 \times 10^{16} \frac{\text{part.}}{\text{sec}}}{6.02 \times 10^{23} \frac{\text{part.}}{\text{mole}}} \times \frac{24.3 \text{ g}}{\text{mole}} \times \frac{3600 \text{ sec}}{\text{hr}} \times \frac{1 \text{ cm}^3 \text{ of Mg}}{1.692 \text{ g}}$$

$$= 1.8 \times 10^{-3} \frac{\text{cm}^3}{\text{hr}} \Rightarrow \frac{8 \text{ cm}^3}{1.8 \times 10^{-3} \frac{\text{cm}^3}{\text{hr}}} = 4446 \text{ hrs of operation}$$

# JET Characteristics

GOMETRY:



Flat distribution cutoff edge  $z_c$

$$\frac{z - \frac{w_1}{a}}{a+b} = \frac{z - \frac{w_2}{a}}{b} \Rightarrow z \left( \frac{1}{b} - \frac{1}{a+b} \right) = \frac{w_2}{ab} - \frac{w_1}{a(a+b)}$$

$$z \cdot \frac{a}{(b+a)b} = \frac{w_2}{ab} - \frac{w_1}{a(a+b)}$$

$$\text{or } z_c = \frac{1}{2a} [(b+a)w_2 - bw_1]$$

Edge  $z_e$

$$\frac{\frac{w_2}{a} + \frac{w_1}{a}}{a} = \frac{z_e + \frac{w_1}{a}}{a+b}$$

$$z_e = \frac{a+b}{2a} (w_2 + w_1) - \frac{w_1}{2}$$

## CORE Jet Density

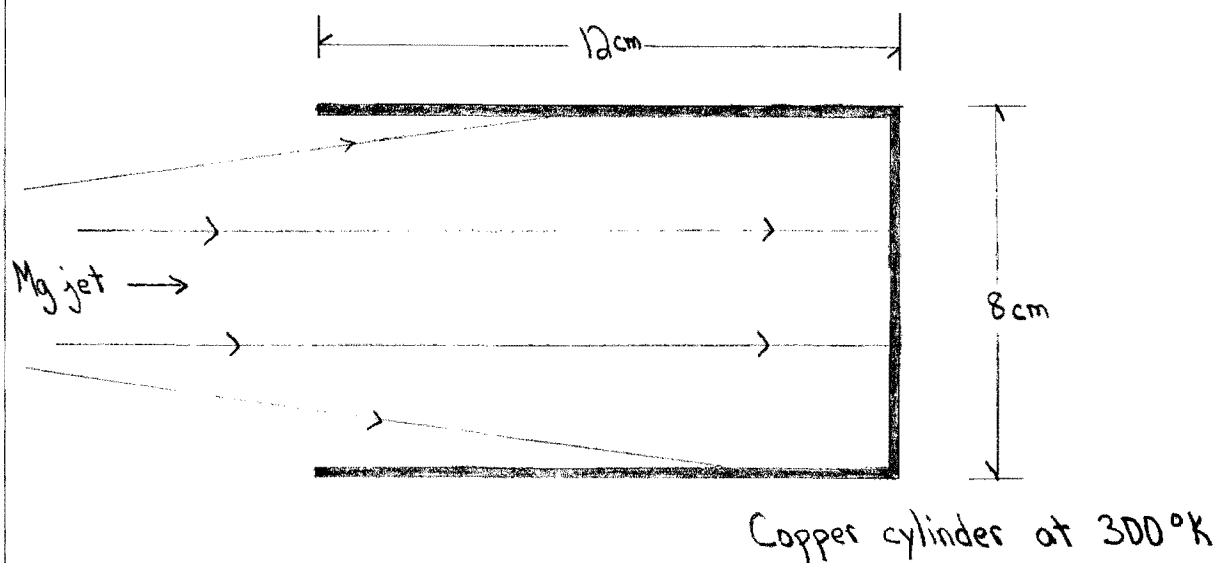
$$\begin{aligned}n_j &= n_0 \Omega = n_0 \frac{A_a}{2\pi d^2} = \frac{(1.47 \times 10^{13} \frac{\text{part}}{\text{cm}^3}) (3 \times 10^{-1} \text{cm}^2)}{2\pi (10 \text{cm})^2} \\ &= 7.02 \times 10^9 \frac{\text{part.}}{\text{cm}^3}\end{aligned}$$

## Ion Count (to be detected by microchannel plates)

$$\begin{aligned}N_c (\text{counts/sec}) &= N_i \sigma_m t (\text{thickness of Mg ribbon}) n_j \\ &= (10^{13} \frac{\text{protons}}{\text{sec}}) (2 \times 10^{-18} \text{cm}^2) (.1 \text{cm}) (7.02 \times 10^9 \frac{\text{part.}}{\text{cm}^3}) \\ &= 14,040 \text{ counts per second}\end{aligned}$$

## COLLECTOR Characteristics

(To ensure the MCP's and beam pipe are not contaminated.)



For Mg impacting on Cu at room temperature, the thermal accommodation coefficient  $\alpha_T = 1$  (i.e. none of the Mg bounces for a jet of 380-400°C). This implies that all the Mg that gets away from the collector does so due to desorption.

$1 - \alpha_s$  ( $\alpha_s \equiv$  sticking coefficient) =  $\alpha_d$ , the amount of Mg which gets away relative to the amount impinging. Assuming a worst case calculation - magnesium spread evenly over collector, maximum evaporation rate (all Mg gets away) - then  $N_e \equiv$  # of particles that get away per second, and is equal to  $uA_3$ .

$u \equiv$  evaporation rate       $A_3 \equiv$  collector cross section  $\approx 74 \text{ cm}^2$

$$u = 3.513 \times 10^{22} \frac{P_v}{(MT)^{1/2}} \text{ sec}^{-1} \text{ cm}^{-2} = 7.127 \times 10^{21} \frac{P_v}{T^{1/2}}$$

$P_v \equiv$  vapor pressure of Mg in Torr at temperature T.

$$\text{where } \log P_v = \frac{-7620}{T} + 2.5 \log T + 1.73736 \quad (\text{for Mg})$$

$$\therefore \text{ at } 300^\circ \text{K} \quad P_v \approx 3.4 \times 10^{-18} \text{ Torr}$$

$$\text{and } u = 1.39 \times 10^3 \text{ part./cm}^2 \cdot \text{sec}$$

$$\text{therefore, } N_e = \left( 1.39 \times 10^3 \frac{\text{part}}{\text{cm}^2 \cdot \text{sec}} \right) \times (74 \text{ cm}^2) = 1.0286 \times 10^5 \frac{\text{part}}{\text{sec}}$$

$$N_w \equiv \text{weight of Mg escaping} = \left( 1.0286 \times 10^5 \frac{\text{part}}{\text{sec}} \right) \times \left( \frac{1 \text{ mole}}{6.02 \times 10^{23} \text{ part}} \right)$$

$$\times \left( 24.3 \frac{\text{g}}{\text{mole}} \right) \times \left( 86,400 \frac{\text{sec}}{\text{day}} \right) \times \left( 365 \frac{\text{days}}{\text{yr}} \right) \approx 1.31 \times 10^{-10} \text{ g/yr}$$

## Lifetime of MCP

It can be assumed that a MCP is ruined by chemical contamination if a monolayer of Mg forms over it. Then assume that during and after baking any Mg in the chamber deposits evenly over everything in  $\sim 1$  meter radius area. Using a spherical surface as a convenience:  $S = 4\pi r^2 \sim 12 \text{ m}^2 \sim 1.2 \times 10^5 \text{ cm}^2$ ,  $D =$  average particle spacing  $\sim 3 \text{ \AA}$ . So, in a monolayer each particle takes up  $\approx$  a surface area  $\Delta S = 7 \times 10^{-16} \text{ cm}^2$ .

$$N \equiv \# \text{ of atoms needed} = \frac{S}{\Delta S} \approx 1.8 \times 10^{20} \text{ atoms} \times 4.04 \times 10^{-23} \frac{\text{g}}{\text{part. of Mg}}$$

$$= 7.2 \times 10^{-3} \text{ g of Mg needed to ruin the plates.}$$

$$\therefore \text{detector lifetime} = T = \frac{7.2 \times 10^{-3} \text{ g}}{1.3 \times 10^{-10} \text{ g/yr}} = 5.55 \times 10^7 \text{ yrs.}$$

Thus, as long as the collector and oven array are separated from the system during baking (see figure 2) there should be no difficulty with plate contamination.

## Sticking Coefficient

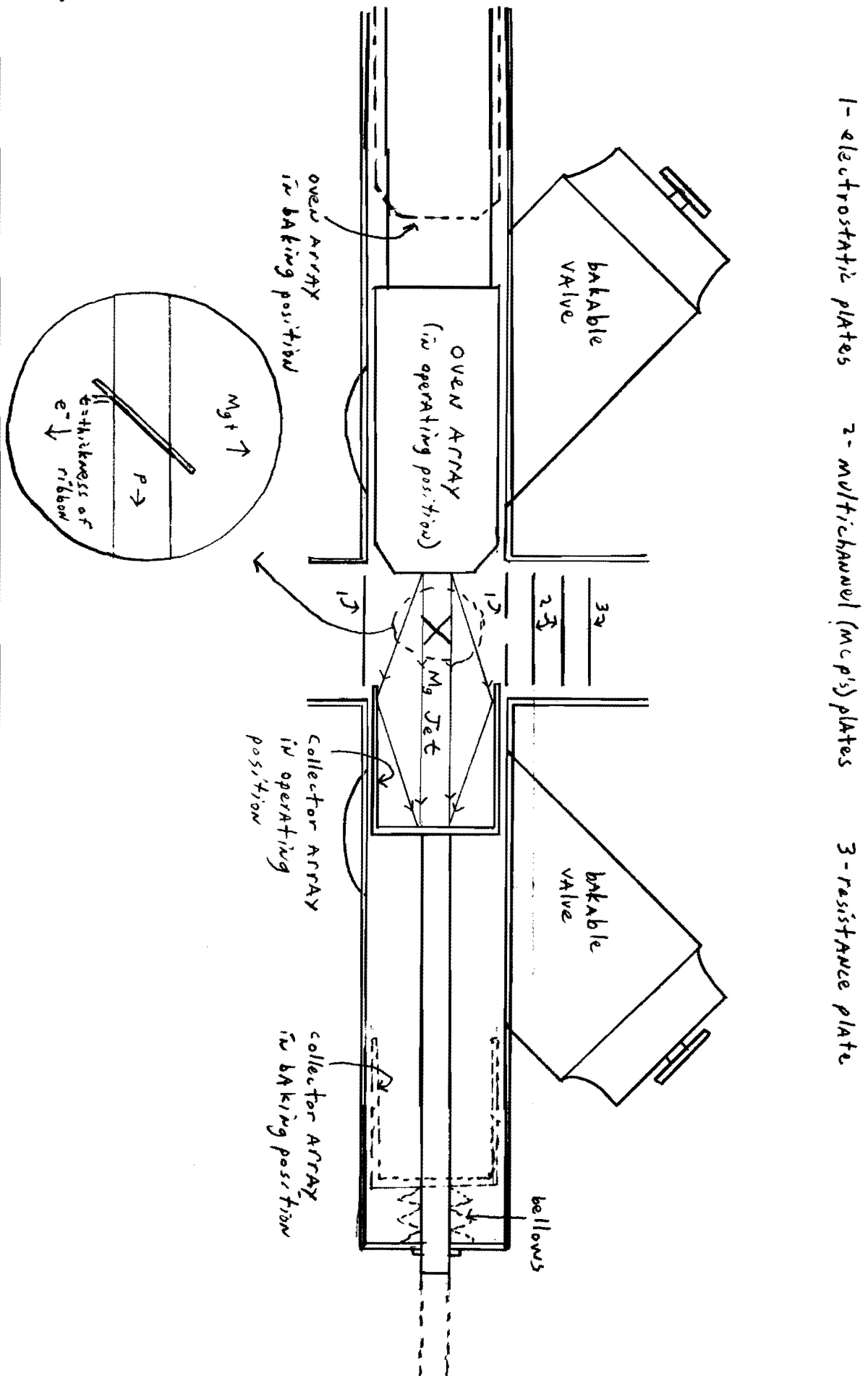
$$r \equiv \text{rate of Mg deposition on collector} = \left( 7.02 \times 10^9 \frac{\text{part}}{\text{cm}^2} \right) \times \left( 7.56 \times 10^4 \frac{\text{cm}}{\text{s}} \right) \times (1.3 \text{ cm}^2)_{\text{area of 2nd slit}} = 1.59 \times 10^{14} \frac{\text{part}}{\text{sec}}$$

$$\left( 1.59 \times 10^{14} \frac{\text{part}}{\text{sec}} \right) \times \left( 1.273 \times 10^{-15} \frac{\text{g} \cdot \text{sec}}{\text{part} \cdot \text{yr}} \right) = .203 \text{ g/yr}$$

$$\text{and } \alpha_d = \frac{N_w}{r} = \frac{1.31 \times 10^{-10} \text{ g/yr}}{.203 \text{ g/yr}} \approx 6.5 \times 10^{-10} \quad \therefore \alpha_s = 1 - \alpha_d \approx 1$$



Figure 2: Complete Setup for the Magnesium Jet



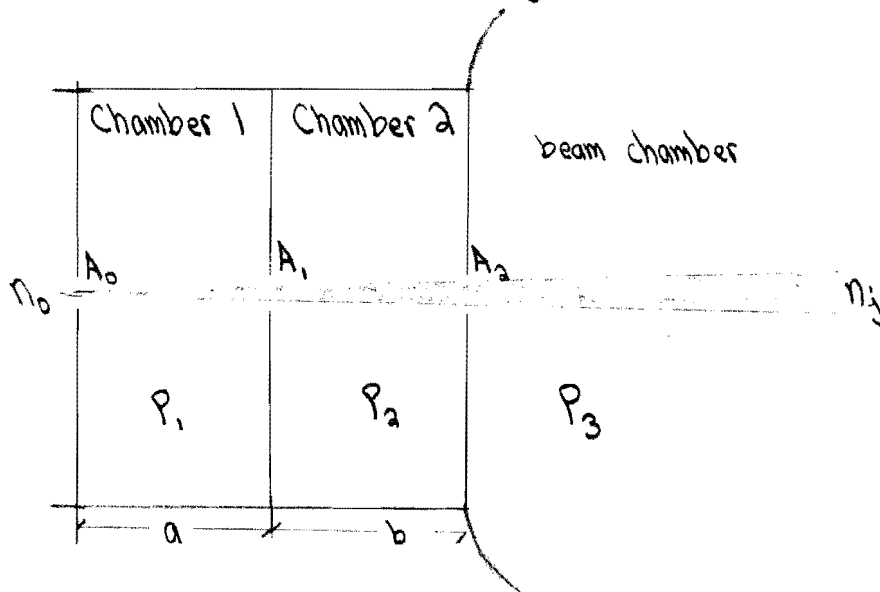
# Hydrogen Jet

## GOALS

- \*  $1 \times 10^4 \frac{\text{ions}}{\text{sec}}$  striking MCP's due to jet
- \* high signal to noise ratio (low background)

## METHOD

Utilize titanium pumping and scrapers to form the jet.



Rough dimensions for calculation purposes:

$$a = 5 \text{ cm}$$

$$b = 5 \text{ cm}$$

$$A_0 = 10^{-2} \text{ cm}^2$$

$$A_1 = 5 \text{ cm} \times .1 \text{ cm}$$

$$A_2 = 5 \text{ cm} \times .1 \text{ cm}$$

## CALCULATIONS

counts wanted from jet  $\sim 1 \times 10^4 \frac{\text{ions}}{\text{sec}}$

$$\text{so, } 1 \times 10^4 \frac{\text{ions}}{\text{sec}} = (\text{Hit probability}) \left( 1 \times 10^7 \frac{\text{protons}}{\text{bunch}} \right) \left( 10^6 \frac{\text{bunch}}{\text{sec}} \right)$$

$$\begin{aligned} &= 1 \times 10^{-9} = n_j \sigma t \\ &= n_j (10^{-18} \text{ cm}^2) (1.10 \text{ cm}) \\ \therefore n_j &= 1 \times 10^{10} \frac{\text{Protons}}{\text{cm}^3} \end{aligned}$$

$$n_j = n_0 \frac{\Delta \Omega}{2\pi}$$
$$\text{so, } n_0 = n_j \frac{2\pi}{\Delta \Omega}$$

$$\Delta \Omega = \frac{A}{r^2} = \frac{(5 \text{ cm})(1.1 \text{ cm})}{100 \text{ cm}^2} = 5 \times 10^{-3}$$

$$\therefore n_0 = (1 \times 10^{10}) \left( \frac{2\pi}{5 \times 10^{-3}} \right) = 1.3 \times 10^{13} \frac{\text{P}}{\text{cm}^3}$$

since  $P = \frac{n}{V} RT$

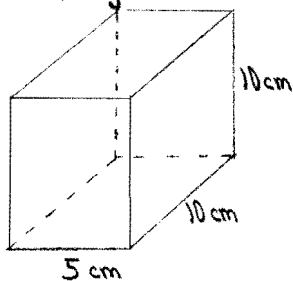
$$\frac{n}{V} = (1.3 \times 10^{13} \frac{\text{P}}{\text{cm}^3}) \left( \frac{\text{mole}}{6.02 \times 10^{23} \text{ part}} \right) = 2.6 \times 10^{-11} \frac{\text{mole}}{\text{cm}^3}$$

$$\begin{aligned} \therefore P_0 &= (2.6 \times 10^{-11} \frac{\text{mole}}{\text{cm}^3}) \left( 6.236 \times 10^4 \frac{\text{cm}^3 \cdot \text{T}}{\text{OK} \cdot \text{mole}} \right) (300 \text{ OK}) \\ &= 4 \times 10^{-4} \text{ Torr} = P_0 \end{aligned}$$

## Titanium Pumping

$10 \frac{\text{g}}{\text{s} \cdot \text{cm}^2}$  maximum for fresh titanium

pumping in Chambers 1 and 2:



$$= 400 \text{ cm}^2$$

more realistic pumping speed

$$(400 \text{ cm}^2) \left( 1 \frac{\text{g}}{\text{s} \cdot \text{cm}^2} \right) = 400 \frac{\text{g}}{\text{s}} = 4 \times 10^5 \frac{\text{cm}^3}{\text{s}}$$

Ultimate pressure in chamber due to gas load:  $P = \frac{Q}{S}$

$Q \equiv$  throughput

$S \equiv$  pumping speed

$$Q_0 = P_0 C_0 \rightarrow \text{conductance}$$

$$= P_0 \frac{\bar{v} A_0}{4} \quad 300^\circ\text{K}$$
$$= (4 \times 10^{-4} \text{ T}) (1.8 \times 10^5 \frac{\text{cm}}{\text{s}}) (1 \times 10^{-2} \text{ cm}^2) \frac{1}{4}$$

$$Q_0 = 1.8 \times 10^{-1} \frac{\text{T cm}^3}{\text{s}} = 1.8 \times 10^{-4} \frac{\text{T} \cdot \text{L}}{\text{s}}$$

$$P_1 = \frac{Q_0}{S} = \frac{1.8 \times 10^{-1} \frac{\text{T cm}^3}{\text{s}}}{4 \times 10^5 \frac{\text{cm}^3}{\text{s}}} = .45 \times 10^{-6} \text{ T} = 4.5 \times 10^{-7} \text{ T}$$

$$Q_1 = P_1 \frac{\bar{v} A_1}{4} = (4.5 \times 10^{-7} \text{ T}) (1.8 \times 10^5 \frac{\text{cm}}{\text{s}}) (.5 \text{ cm}^2) \frac{1}{4}$$
$$= .010 \frac{\text{cm}^3 \cdot \text{T}}{\text{s}}$$

$$P_2 = \frac{Q_1}{S} = \frac{.010 \frac{\text{cm}^3 \cdot \text{T}}{\text{s}}}{4 \times 10^5 \frac{\text{cm}^3}{\text{s}}} = 2.5 \times 10^{-8} \text{ T}$$

$$Q_2 = P_2 \frac{\bar{v} A_2}{4} = (2.5 \times 10^{-8} \text{ T}) \left( \frac{1.8 \times 10^5 \text{ cm}}{\text{s}} \right) (.5 \text{ cm}^2)$$
$$= 5.6 \times 10^{-4} \frac{\text{cm}^3 \cdot \text{T}}{\text{s}}$$

$$P_3 = \frac{Q_2}{S} = \frac{5.6 \times 10^{-4} \frac{\text{cm}^3 \cdot \text{T}}{\text{s}}}{4 \times 10^5 \frac{\text{cm}^3}{\text{s}}} = 1.4 \times 10^{-9} \text{ T}$$

jet counts: 10,000 cps

background:

$$n_b \sigma t (1 \times 10^{13} \frac{\text{protons}}{\text{sec}})$$

$$t = 3.5 \text{ cm (region in beam pipe)}$$

$$n_b = \frac{P(T)}{(3.12 \times 10^{-17} \frac{T \text{ cm}^3}{\text{Part}})}$$

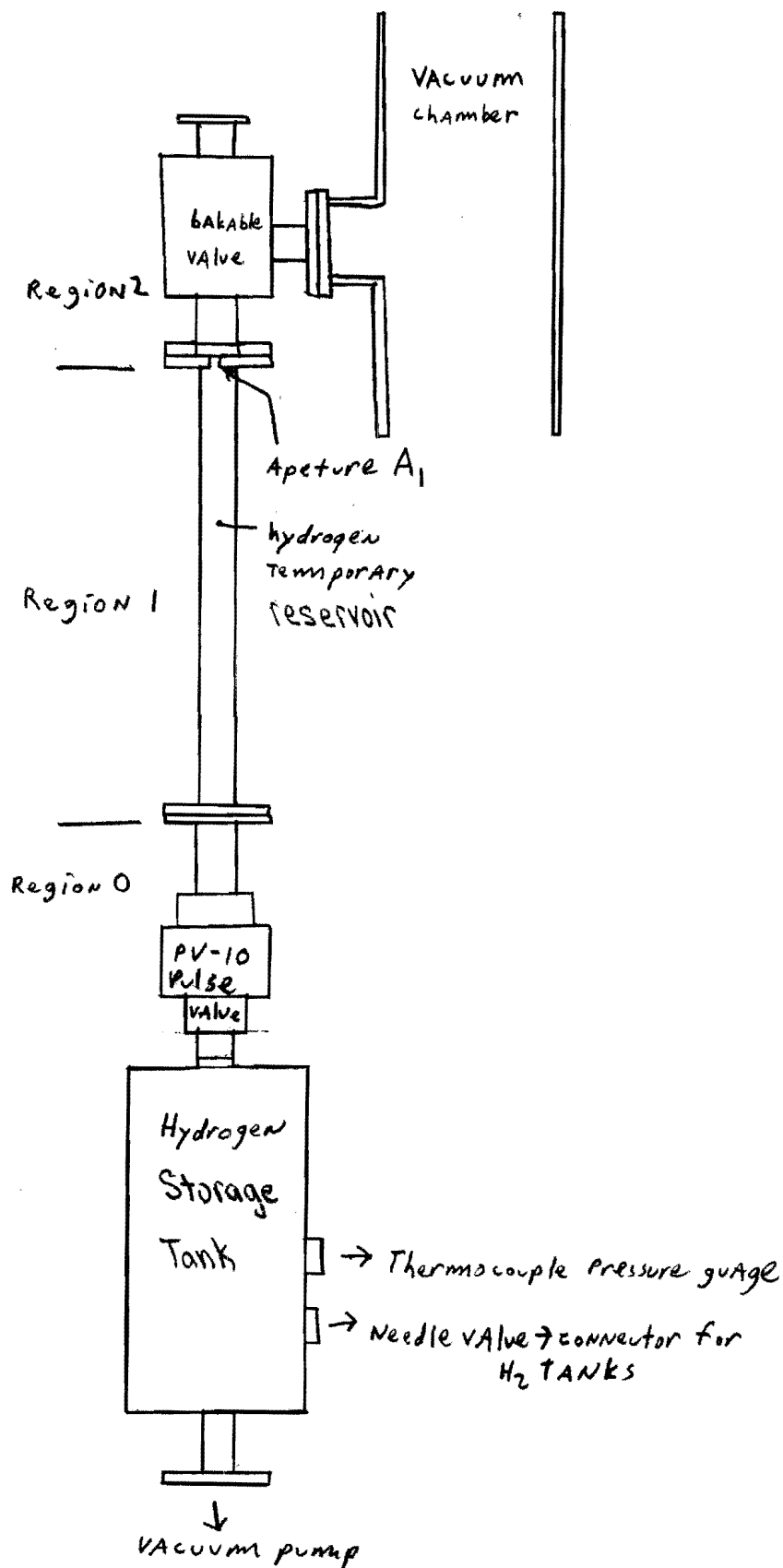
at 300 °K

$$n_b = \frac{1.4 \times 10^{-9} T}{3.12 \times 10^{-17}} = 4.49 \times 10^7 \frac{\text{p}}{\text{cm}^3}$$

$$(4.49 \times 10^7 \frac{\text{p}}{\text{cm}^3}) (1 \times 10^{-18} \text{ cm}^2) (3.5 \text{ cm}) (1 \times 10^{13} \frac{\text{p}}{\text{s}}) = 1,571 \text{ cps}$$

$$S/N \sim 6$$

Figure III: The Hydrogen bump



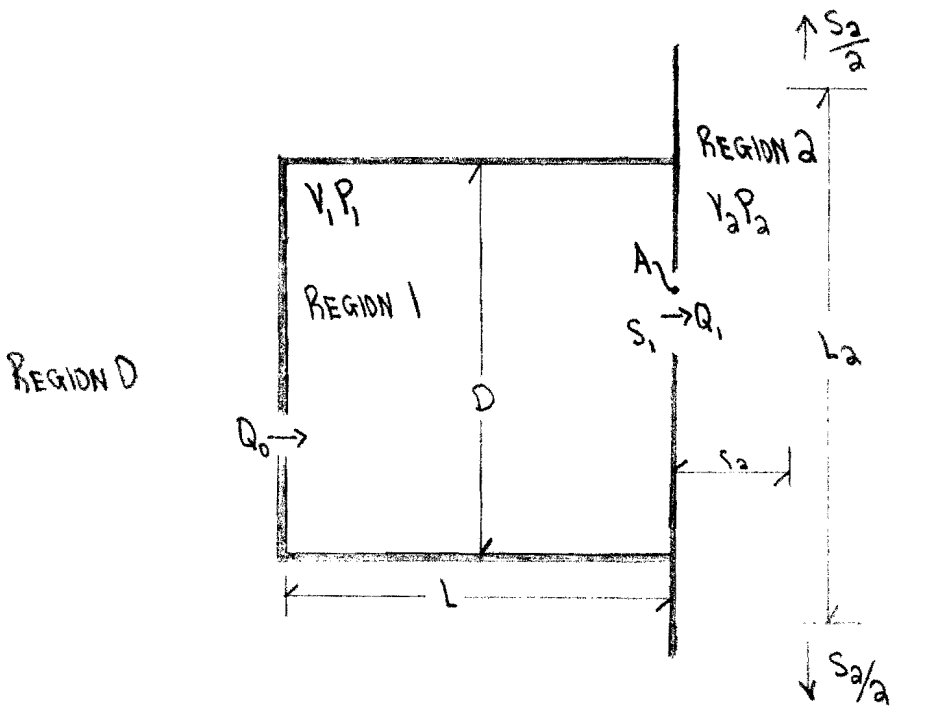
# Hydrogen Bump

## Beam Characteristics

$$10^7 \text{ protons at } 1 \text{ MHz} \Rightarrow N_p = 10^{13} \text{ p/sec}$$

$$\sigma_i \equiv \text{cross section for hydrogen ionization} \sim 10^{-18} \text{ cm}^2$$

## Bump Geometry and Characteristics



$$A \equiv \text{aperture area} = \pi \frac{d_A^2}{4}$$

$$V_2 \equiv \pi r_2^2 L_2 = \pi (5 \text{ cm})^2 (100 \text{ cm}) = 7.854 \text{ l}$$

$$V_1 = \pi \frac{D^2}{4} L$$

$P_i \equiv$  pressure in region  $i$ ; in Torr

$Q_i \equiv$  throughput from region  $i$  to region  $i+1$ , in  $\frac{\text{T}\cdot\text{l}}{\text{sec}}$

$t \equiv$  time in seconds

$S_i \equiv$  pumping speed; in  $\text{l/s}$

$$S_2 = 600 \text{ l/s}$$

$$T \equiv 300 \text{ K}$$

$$m \equiv 2.016 \text{ g/mole } (H_2)$$

$c_i \equiv$  conductance of aperture  $i$

Derivations      time region 1

$$t < 5 \text{ ms}$$

assumptions:

$$Q_0 > 0$$

$$P_1 \gg P_2$$

$$Q_0 \gg P_1 S_1$$

$$d_A \ll D$$

$$S_1 = C_A \left[ 1 - \frac{P_2}{P_1} \right] \approx C_A$$

$$Q_0 - P_1 S_1 = V_1 \frac{dP_1}{dt} \approx Q_0$$

$$\Rightarrow P_1 = \frac{Q_0 t}{V_1}$$

$$Q_1 = P_1 S_1 \sim P_1 C_A$$

$$Q_1 - P_2 S_2 = V_2 \frac{dP_2}{dt} \Rightarrow \frac{Q_0 C_A}{V_1 V_2} t - \frac{P_2 S_2}{V_2} = \frac{dP_2}{dt}$$

Let  $P_2 = a + bt + ce^{dt}$

then,

$$\frac{Q_0 C_A}{V_1 V_2} t - \frac{(a + bt + ce^{dt}) S_2}{V_2} = b + ce^{dt}$$

implying that:  $d = -\frac{S_2}{V_2}$        $a = -\frac{V_2}{S_2} b$        $b = \frac{Q_0 C_A}{V_1 S_2}$

$$a = \frac{-Q_0 C_A V_2}{V_1 S_2^2}$$



$$\text{so, } P_a = \frac{-Q_0 C_A V_2}{V_1 S_a^2} + \frac{Q_0 C_A}{V_1 S_a} t + C e^{-s_a t / V_2}$$

$$\text{let } t=0, \text{ then } P_a = P_{ai} = \frac{-Q_0 C_A V_2}{V_1 S_a^2} + C$$

$$\text{or } C = P_{ai} + \frac{Q_0 C_A V_2}{V_1 S_a^2}$$

$$\text{finally: } P_a = \frac{-Q_0 C_A V_2}{V_1 S_a^2} + \frac{Q_0 C_A}{V_1 S_a} t + \left( P_{ai} + \frac{Q_0 C_A V_2}{V_1 S_a^2} \right) e^{-s_a t / V_2}$$

at  $\Delta t$  (5 msec):

$$P_1 = P_{11} = \frac{Q_0}{V_1} \Delta t$$

$$P_a = P_{a2} = \frac{-Q_0 C_A V_2}{V_1 S_a^2} + \frac{Q_0 C_A}{V_1 S_a} \Delta t + \left( P_{ai} + \frac{Q_0 C_A V_2}{V_1 S_a^2} \right) e^{-\frac{s_a \Delta t}{V_2}}$$

Time Region 2

$t > 5 \text{ msec}$

assumptions:

$$Q_0 = 0$$

$$P_1 \gg P_a$$

$$S_a V_1 \neq C_A V_2$$

$$S_1 \approx C_A$$

$$-P_1 S_1 = V_1 \frac{dP_1}{dt}$$

$$P_1 = P_{11} e^{-\left(\frac{S_1 t}{V_1}\right)} = P_{11} e^{-\frac{C_A t}{V_1}}$$

$$Q_1 = P_1 S_1 \approx P_1 C_A \approx P_{11} C_A \exp\left[-\frac{C_A t}{V_1}\right]$$

$$Q_1 - P_a S_a = V_2 \frac{dP_a}{dt} \Rightarrow \frac{P_{11} C_A}{V_1} \exp\left[-\frac{C_A t}{V_1}\right] - \frac{P_a S_a}{V_2} = \frac{dP_a}{dt}$$

## H<sub>2</sub> Depletion in Storage Tank:

Assuming that the tank is refilled when its pressure has dropped to .90 P<sub>0</sub> (i.e. .1 P<sub>0</sub> V<sub>T</sub> units of gas utilized), and the amount of H<sub>2</sub> used per shot is Q<sub>0</sub> Δt, the number of shots obtained before refilling is:

$$N = \frac{P_0 V_T}{10 Q_0 \Delta t}$$

## Pressure in Beam Pipe if Valve Fails Open

Here the pressure in P<sub>1</sub> will increase until

$$Q_1 = Q_0 \Rightarrow P_2 = \frac{Q_0}{S_2}$$

## Number of Counts Obtained

$$\text{H}_2 \text{ pressure: } P(\text{Torr}) = \frac{n_{\text{H}_2} \left( \frac{\text{part}}{\text{cm}^3} \right) \times R_0 \times T(\text{K})}{N_A} = \frac{n R_0 T}{N_A}$$

$$\text{using } R_0 = 6.263 \times 10^4 \frac{\text{Torr} \cdot \text{cm}^3}{\text{mole} \cdot \text{K}}$$

$$T = 300^\circ \text{K}$$

$$\Rightarrow n_{\text{H}_2} = (3.2 \times 10^{16}) P(\text{Torr})$$

$$N_i \equiv \# \text{ of counts/sec} = n_{\text{H}_2} \sigma_i L_1 N_p$$

$$L_1 \equiv \text{effective MCP length} \\ = 10 \text{ cm}$$

$$\text{and } N_i = (3.2 \times 10^{16}) (10^{-18} \text{ cm}^2) (10 \text{ cm}) (10^{13} \frac{\text{part}}{\text{sec}}) P(\text{Torr}) \\ = (3.2 \times 10^{12}) P(\text{Torr}) \frac{\text{counts}}{\text{sec}}$$

implying that

$$3.125 \times 10^{-13} \text{ T.s of gas yields 1 count}$$
$$1 \text{ count} \equiv 3.125 \times 10^{-13} \text{ T.s}$$

## Applied Design

$$L = 30 \text{ cm}$$

$$D = 1.27 \text{ cm}$$

$$D_A = .15 \text{ cm}$$

$$C_A = 2.86 \left(\frac{T}{M}\right)^{1/2} D_A^2 \frac{1}{\text{sec}} = .785 \frac{\text{g}}{\text{sec}}$$

$$T = 300^\circ \text{K}$$

$$V_2 = 7.84 \text{ l}$$

$$S_2 = 600 \frac{\text{g}}{\text{sec}}$$

$$V_T \sim 1 \text{ l}$$

$$\Delta t = 5 \times 10^{-3} \text{ sec}$$

$$V_1 = .038 \text{ l}$$

$$\frac{C_A}{V_1} = 20.66$$

$$\frac{S_2}{V_2} = 76.53$$

$$P_{2i} = 5 \times 10^{-10} \text{ Torr}$$

$$S_2 V_1 - C_A V_2 = 16.646 \frac{\text{g}^2}{\text{sec}}$$

$$C_A V_1 = 2.98 \times 10^{-2} \frac{\text{g}^2}{\text{sec}}$$

$$\frac{C_A V_1}{S_2 V_1 - C_A V_2} = 1.79 \times 10^{-3}$$

$$\frac{C_A V_2}{V_1 S_2} = 4.5 \times 10^{-4}$$

$$\frac{C_A}{V_1 S_2} = 3.443 \times 10^{-2}$$

For the PV-10 pulse valve,  $Q_0 = C_1 P_0$ , where  $C_1$  varies depending on the applied signal. We use an average  $C_1$  value of  $2.6 \times 10^{-3} \frac{\text{g}}{\text{sec}}$ .

$$\text{Let } P_2 = a e^{-\frac{s_2 t}{V_2}} + b e^{-\frac{c t}{V_1}}$$

then,

$$\begin{aligned} \frac{P_{11} C_A}{V_2} e^{-\frac{c t}{V_1}} - \frac{S_2}{V_2} a e^{-\frac{s_2 t}{V_2}} - \frac{S_2}{V_2} b e^{-\frac{c t}{V_1}} \\ = -\frac{a S_2}{V_2} e^{-\frac{s_2 t}{V_2}} - \frac{b C_A}{V_1} e^{-\frac{c t}{V_1}} \end{aligned}$$

$$\text{so, } \frac{P_{11} C_A}{V_2} - \frac{S_2}{V_2} b + \frac{C_A b}{V_1} = 0$$

$$\Rightarrow b \left( \frac{C_A}{V_1} - \frac{S_2}{V_2} \right) = -\frac{P_{11} C_A}{V_2}$$

$$\Rightarrow b = \frac{-P_{11} C_A V_1}{C_A V_2 - S_2 V_1} = \frac{P_{11} C_A V_1}{S_2 V_1 - C_A V_2}$$

$$\text{and } P_2 = a e^{-\frac{s_2 t}{V_2}} + \frac{P_{11} C_A V_1}{S_2 V_1 - C_A V_2} e^{-\frac{c t}{V_1}}$$

$$\text{at } t=0 \quad P_2 = P_{22} = a + \frac{P_{11} C_A V_1}{S_2 V_1 - C_A V_2}$$

$$\text{or, } a = P_{22} - \frac{P_{11} C_A V_1}{S_2 V_1 - C_A V_2}$$

finally,

$$P_2 = \frac{P_{11} C_A V_1}{S_2 V_1 - C_A V_2} e^{-\frac{c t}{V_1}} + \left( P_{22} - \frac{P_{11} C_A V_1}{S_2 V_1 - C_A V_2} \right) e^{-\frac{s_2 t}{V_2}}$$

time region 1

$$P_1 = (6.92 \times 10^{-2}) P_0 t$$

$$P_2 = (-1.183 \times 10^{-6}) P_0 + (9.06 \times 10^{-5}) P_0 t + (5 \times 10^{-10} + 1.183 \times 10^{-6} P_0) e^{-76.5t}$$

$$P_{11} = 3.46 \times 10^{-4} P_0$$

$$P_{22} = (-7.3 \times 10^{-7}) P_0 + (1.682)(5 \times 10^{-10} + 1.183 \times 10^{-6} P_0)$$

time region 2

$$P_1 = P_{11} e^{-20.66t}$$

$$P_2 = P_{11} (1.792 \times 10^{-3}) e^{-20.66t} + (P_{22} - P_{11} (1.792 \times 10^{-3})) e^{-76.53t}$$

depletion in storage tank

$$N = \frac{P_0 (12)}{10 \left( 2.63 \times 10^{-3} \frac{\text{g}}{\text{sec}} \right) P_0 (5 \times 10^{-3} \text{sec})} \approx 76.05 \text{ shots}$$

valve failure pressure

$$P = \frac{2.63 \times 10^{-3} P_0}{600 \frac{\text{g}}{\text{sec}}} = 4.38 \times 10^{-6} P_0$$

Results

Choosing  $P_0 = 2 \times 10^{-2} \text{ Torr}$

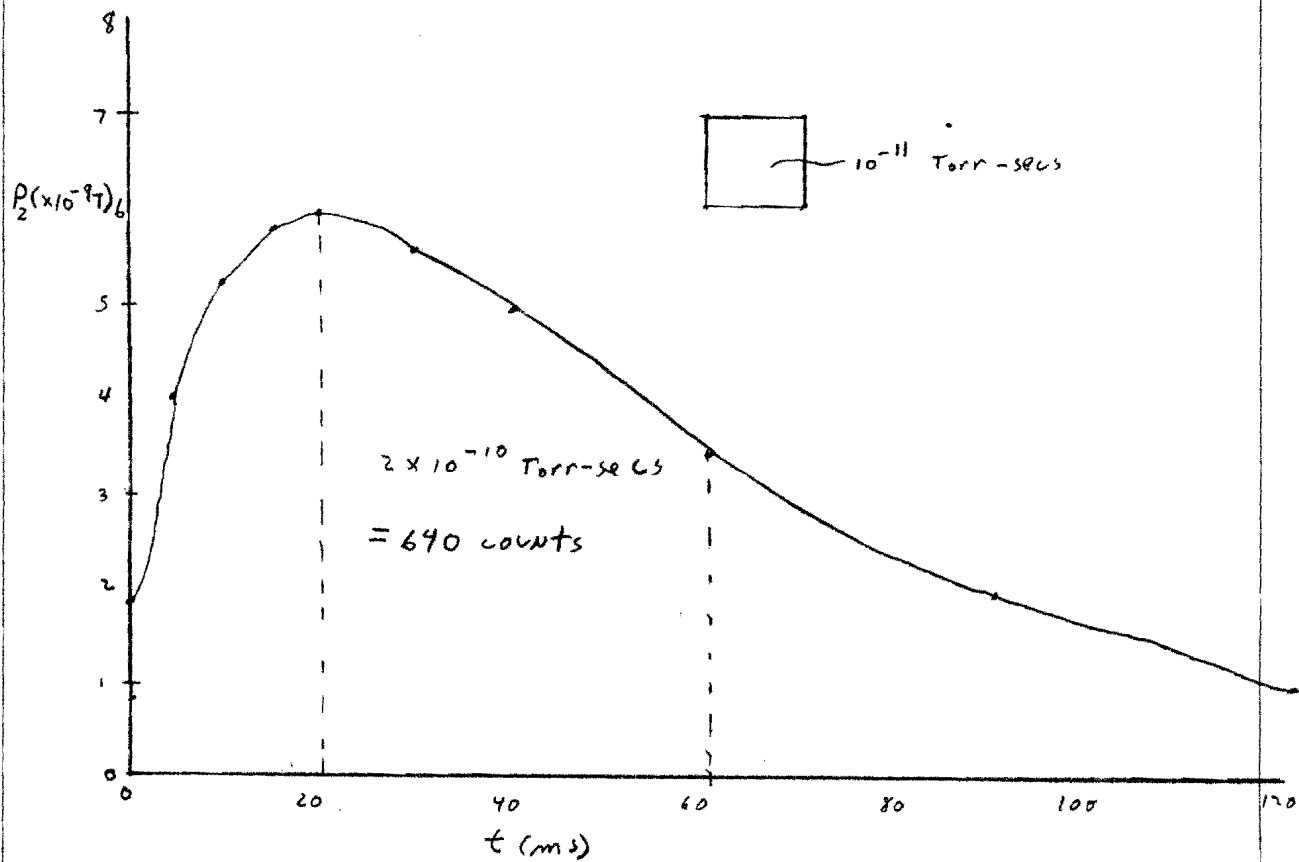
then,  $P_{11} = 6.92 \times 10^{-6} \text{ T}$

$$P_{22} = 1.88 \times 10^{-9} \text{ T}$$

$$P_2 = (1.24 \times 10^{-8}) e^{-20.66t} - (1.052 \times 10^{-8}) e^{-76.53t}$$

$$P_{\text{fail}} = 8.76 \times 10^{-8} \text{ T}$$

$t$ (beyond initial 5 ms)	real time (ms)	$P_2$ ( $\times 10^{-9}$ T)
0 ms	5	1.88
5	10	4.01
10	15	5.19
15	20	5.76
20	25	5.93
30	35	5.61
40	45	4.93
60	65	3.48
90	95	1.92
120	125	1.04



Designs prepared in coordination with:

Fred Mills, Fermilab

Igor Meshkov, Institute of Nuclear Physics, U.S.S.R.

Bill Kells, Fermilab

Don Young, Fermilab

Dave Cline, University of Wisconsin

Tom Hardek, Fermilab